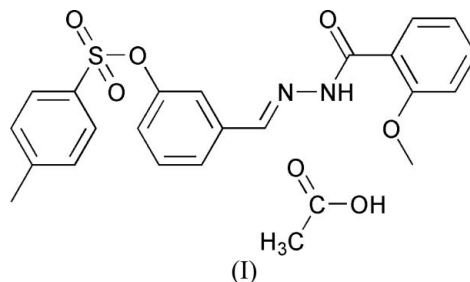


(E)-2-Methoxy-N'-[3-(4-methylphenylsulfonyloxy)-benzylidene]benzohydrazide acetic acid solvate**Xin Chen* and Ming Yu**College of Sciences, Tianjin University of
Science and Technology, Tianjin 300222,
People's Republic of ChinaCorrespondence e-mail:
chen_xin9999@163.com**Key indicators**Single-crystal X-ray study
 $T = 294$ K
Mean $\sigma(\text{C}-\text{C}) = 0.004$ Å
 R factor = 0.049
 wR factor = 0.154
Data-to-parameter ratio = 13.3For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

In the title compound, $\text{C}_{22}\text{H}_{20}\text{N}_2\text{O}_5\text{S}\cdot\text{C}_2\text{H}_4\text{O}_2$, the central benzene ring makes dihedral angles of 55.41 (9) and 3.88 (14) $^\circ$ with the two other benzene rings. An intramolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bond helps to stabilize the molecular conformation, while intermolecular $\text{O}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds link adjacent molecules to form an infinite network.

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Schiff base ligands have received a good deal of attention in biology and chemistry (Kahwa *et al.*, 1986). Many Schiff base derivatives have been synthesized and employed to develop protein and enzyme mimics (Santos *et al.*, 2001), such as models to mimic hydrolase in the hydrolysis of *p*-nitrophenyl picolinate (Li *et al.*, 2005). Structural information is useful when investigating the coordination properties of Schiff bases functioning as ligands. We report here the synthesis and molecular structure of the title Schiff base compound, (I) (Fig. 1).



In (I), bond lengths and angles are within normal ranges (Allen *et al.*, 1987). The central benzene ring (atoms C8–C14/O3) makes dihedral angles of 55.41 (9) and 3.88 (14) $^\circ$, respectively, with the C1–C6 and C16–C21 benzene rings, while the dihedral angle between these two benzene rings is 59.15 (10) $^\circ$.

An intramolecular hydrogen bond links the NH group to atom O5, thereby influencing the molecular conformation. There is also an $\text{O}-\text{H}\cdots\text{O}$ intermolecular hydrogen bond linking the main molecule and the solvent molecule. The crystal structure is stabilized by three weak non-classical intermolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds (Table 1), forming an infinite network (Fig. 2).

Experimental

An anhydrous ethanol solution (50 ml) of 3-formylphenyl-4-methylbenzenesulfonic acid (2.76 g, 10 mmol) was added to an anhydrous ethanol solution (50 ml) of 2-methoxybenzohydrazide (1.66 g,

10 mmol) and the mixture stirred at 350 K for 3 h under N₂, giving a white precipitate. The product was isolated, recrystallized from ethanol, and then dried in a vacuum to give pure compound (I) in 85% yield. Colorless single crystals of (I) suitable for X-ray analysis were obtained by slow evaporation of an ethanol–acetic acid (1:1) solution.

Crystal data

C ₂₂ H ₂₀ N ₂ O ₅ S·C ₂ H ₄ O ₂	$V = 1175.7(4) \text{ \AA}^3$
$M_r = 484.52$	$Z = 2$
Triclinic, $P\bar{1}$	$D_x = 1.369 \text{ Mg m}^{-3}$
$a = 8.2869(15) \text{ \AA}$	Mo $K\alpha$ radiation
$b = 8.9350(16) \text{ \AA}$	$\mu = 0.19 \text{ mm}^{-1}$
$c = 16.375(3) \text{ \AA}$	$T = 294(2) \text{ K}$
$\alpha = 92.368(3)^\circ$	Block, colorless
$\beta = 102.547(3)^\circ$	$0.26 \times 0.24 \times 0.20 \text{ mm}$
$\gamma = 95.484(3)^\circ$	

Data collection

Bruker SMART APEX CCD area-detector diffractometer	6010 measured reflections
φ and ω scans	4118 independent reflections
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	3137 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.938$, $T_{\max} = 0.964$	$R_{\text{int}} = 0.015$
	$\theta_{\text{max}} = 25.0^\circ$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0825P)^2 + 0.3853P]$
$R[F^2 > 2\sigma(F^2)] = 0.049$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.154$	$(\Delta/\sigma)_{\text{max}} = 0.002$
$S = 1.07$	$\Delta\rho_{\text{max}} = 0.55 \text{ e \AA}^{-3}$
4118 reflections	$\Delta\rho_{\text{min}} = -0.35 \text{ e \AA}^{-3}$
310 parameters	
H-atom parameters constrained	

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N2–H2 \cdots O5	0.86	1.93	2.616 (3)	135
O6–H6A \cdots O4 ⁱ	0.82	2.06	2.709 (3)	136
C10–H10 \cdots O1 ⁱ	0.93	2.49	3.239 (3)	137
C6–H6 \cdots O1 ⁱⁱ	0.93	2.56	3.382 (4)	148
C7–H7A \cdots O1 ⁱⁱⁱ	0.96	2.54	3.460 (4)	160

Symmetry codes: (i) $x, y + 1, z$; (ii) $-x + 1, -y, -z + 1$; (iii) $-x, -y, -z + 1$.

The H atoms were included in calculated positions and refined using a riding-model approximation. Constrained C–H and N–H bond lengths and isotropic $U_{\text{iso}}(\text{H})$ parameters: 0.93 \AA and $1.2U_{\text{eq}}(\text{C})$ for Csp², 0.96 \AA and $1.5U_{\text{eq}}(\text{C})$ for methyl, and 0.86 \AA and $1.2U_{\text{eq}}(\text{N})$ for imino H atoms.

Data collection: SMART (Bruker, 1999); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997a); molecular graphics: SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: SHELXTL.

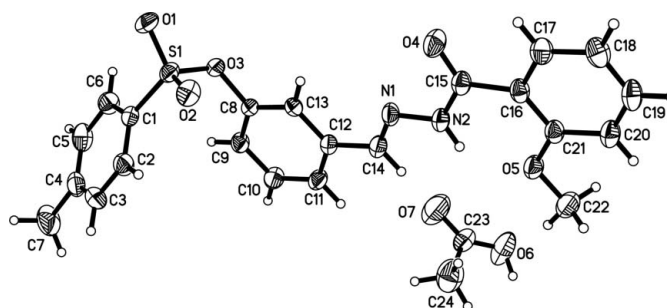


Figure 1

The molecular structure of (I), showing the atom-labeling scheme, with displacement ellipsoids for non-H atoms drawn at the 30% probability level.

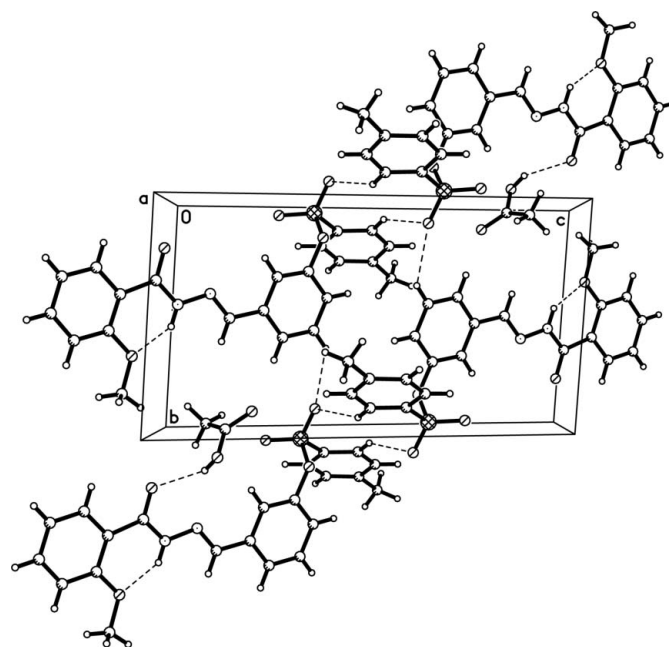


Figure 2

A packing diagram for (I), with hydrogen bonds shown as dashed lines.

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